

bonds, viz. one out of the 3 possible in α -residues, three out of the 6 possible in β - and γ -residues, and two out of the 3 possible in δ -residues (cf. structure 2).

This at least suggests that reduction of the postulated precursor tri-*O*-geranylgeranylglycerol occurs progressively from the oxygenated end of the isoprenoid chains. Unfortunately the structures of these very minor triethers do very little to clarify the biogenesis of the ω,ω' -bi-phytanyl skeleton in the major cyclic diether lipids of *Caldariella*.

EXPERIMENTAL

Culture methods and lipid extraction procedures are described elsewhere [1,3]. Direct extraction of the total lipid (25 g) with hexane gave a soluble fraction (540 mg) which was subjected to chromatography on Merck Kieselgel (70–230 mesh) in hexane with increasing proportions of Et₂O. The triethers eluted with 20% Et₂O and were further separated into (1) and (2) by TLC (Merck Kieselgel 60-F254) in hexane-ether (7:3) (*R_f* 0.5 and 0.4 respectively).

Hydrogenation of (2). The triether (10 mg) was treated with H₂ on Pd-C (5%; 15 mg) in MeOH (3 ml) for 24 hr; chromatography of the product, as above, gave (1) (NMR, IR) which was subsequently converted as for the natural saturated triether.

Ether cleavage etc. Treatment of (1) (15 mg) with 57% HI (1 ml; 24 h reflux) and work-up (extraction into hexane, washing with H₂O, aq. K₂CO₃, aq. Na₂S₂O₃, and H₂O, partition with hexane and 90% aq. MeOH, and TLC in hexane) gave

phytanyl iodide, C₂₀H₄₁I (MS and NMR). The product (10 mg) was refluxed with AgOAc (20 mg) in AcOH (2 ml) for 24 hr. The reaction mixture was taken up in Et₂O, centrifuged, washed (H₂O, aq. NaCl, aq. NaHCO₃, aq. Na₂S₂O₃, H₂O) and purified by TLC in hexane-ether (17:3). The product *R_f* 0.75, was identified as phytanyl acetate by comparison with authentic material [MS, NMR, and GLC (on a 2m × 3mm glass column packed with 1% OV-1 on Gas-Chrom Q 100/120 with 40 ml/min N₂ at 200°; retention time 6.24 min)]. Alternatively, the triether (10 mg) was treated with BCl₃ (1 ml with 1 ml CHCl₃; 14 hr at R.T.). Excess reagent was removed at reduced pressure and the alkyl chloride taken up in hexane for purification by TLC as above (phytanyl chloride, C₂₀H₄₁Cl by MS and NMR). The undissolved residue was taken up in methanol and identified as glycerol by TLC (in CHCl₃-MeOH, 4:1 on Si gel, *R_f* 0.40, visualised with Ag-NH₃).

Spectra. All IR spectra were measured in CCl₄; NMR spectra at 100 MHz in CCl₄ with TMS standard; MS on the AEI MS-30.

REFERENCES

1. de Rosa, M., Gambacorta, A. and Bu'Lock, J. D. (1975) *J. Gen. Microbiol.* **86**, 156.
2. de Rosa, M., Gambacorta, A., Minale, L. and Bu'Lock, J. D. (1974) *Chem. Commun.* 543.
3. de Rosa, M., Gambacorta, A. and Bu'Lock, J. D. (1976) *Phytochemistry* **15**, 143.
4. de Rosa, M., Gambacorta, A. and Minale, L. (1975) *Chem. Commun.* 392.

Phytochemistry, 1976, Vol. 15, pp 1996–1997. Pergamon Press Printed in England

6 β ,22-DIHYDROXYHOPANE, A NEW TRITERPENE FROM THE FERN *CHEILANTHES MARANTAE*

ANTONIO G. GONZALEZ, CARMEN BETANCOR, ROSENDO HERNANDEZ and JOSE A. SALAZAR

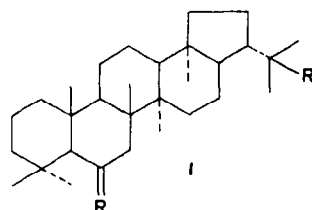
Department of Organic and Biochemistry, University of La Laguna. Instituto de Productos Naturales Orgánicos, CSIC, Tenerife, Spain

(Received 31 May 1976)

Key Word Index—*Cheilanthes marantae*; Sinopteridaceae; doradilla acanalada; fern; triterpenes, steroids; diplopterol; fernenol; 6 β ,22-dihydroxyhopane; sitosterol; sitosterol-*o*- β -D-glucoside.

Plant. *Cheilanthes marantae* L. local name doradilla acanalada. Specimen no. 322 deposited in the Herbarium of the Botany Department, University of La Laguna. **Source.** Monte de las Mercedes, Tenerife, Canary Islands. Collected in June. **Previous work.** None.

Present work. In addition to saccharose, diplopterol, fernenol, sitosterol and sitosterol-*o*- β -D-glucoside, the new natural triperpene 6 β ,22-dihydroxyhopane (**1a**), characterized by its IR, PMR and MS spectra, was isolated from the stems and leaves of *Cheilanthes marantae*. The hopane framework of (**1a**) was deduced from its MS which showed the same fragmentation pattern as that of zeorin (**1d**) and related compounds [1]. The large downfield shifts of the methyl groups at C-4 β , C-8 β and C-10 β observed in the PMR spectrum (ca 0.35 ppm) compared with 22-hydroxyhopane are compatible only with the presence of a C-6 β hydroxyl group [2]. These assumptions were chemically confirmed by partial synthesis of (**1a**) from zeorin.



- | | | |
|-----------|---------------------|----------|
| 1a | R = H; β -OH | R' = OH |
| 1b | R = H; β -OAc | R' = OAc |
| 1c | R = O | R' = OH |
| 1d | R = H; α -OH | R' = OH |

EXPERIMENTAL

Mp's determined on a Koffler hot-stage apparatus, are uncorr. Optical rotations were measured in CHCl₃ and PMR spectra in CDCl₃ with TMS as internal reference.

Extraction. Leaves and stems were extracted with hot EtOH and conc in *vacuo*. Saccharose was separated by crystallization from the concentrate and the residue was chromatographed on Si gel and eluted with C_6H_6 , C_6H_6 -AcOEt, AcOEt, giving: diplopterol ($2.7 \times 10^{-3}\%$ of plant dry wt.), fernenol ($3.4 \times 10^{-3}\%$), sitosterol, the new natural triterpene 6 β ,22-dihydroxyhopane (**1a**, $4.8 \times 10^{-4}\%$) and sitosterol- β -D-glucoside.

Diplopterol. Mp 234–236° (MeOH–Me₂CO), $[\alpha]_D + 42^\circ$ (c 0.262). IR ν_{max}^{KBr} cm^{-1} : 3450 (OH). PMR δ : 0.79 (3H, s, C-28), 0.84 (6H, s, C-23, C-24), 0.86 (3H, s, C-25), 0.98 (6H, s, C-26, C-27), 1.21 (6H, s, C-29, C-30). MS (70eV), m/e (rel. int.): 428 (M^+ , 5), 207 (21), 191 (59), 189 (49), 149 (54), 121 (21), 95 (71), 59 (100).

Fernenol. Mp 198–200° (MeOH) $[\alpha]_D - 23^\circ$ (c 0.216) [3]. IR ν_{max}^{KBr} cm^{-1} : 3610 (OH). PMR δ : 0.72, 0.76, 0.83, 0.88, 0.95, 1.06 (24H, eight Me groups), 3.15 (1H, m, $W_{1/2} = 18$ Hz, C-3), 5.35 (1H, m, $W_{1/2} = 10$ Hz, C-11). MS (70eV), m/e (rel. int.): 426 (M^+ , 50), 411 (100), 393 (20), 273 (12), 255 (12), 259 (94), 241 (30), 95 (34). **Acetate.** Mp 218° (Me₂CO), $[\alpha]_D - 6^\circ$ (c 0.246). (Found: C, 81.61; H, 11.15. Calc. for $C_{32}H_{52}O_3$: C, 81.99; H, 11.18%.) On chromic acid oxidation, fernenol furnished the corresponding ketone. Mp 197° (MeOH), $[\alpha]_D - 47^\circ$ (c 0.254). IR $\nu_{max}^{CHCl_3}$ cm^{-1} : 1700 (CO). MS (15eV), m/e (rel. int.): 424 (M^+ , 18), 410 (12), 409 (34), 271 (20), 258 (20), 257 (100), 246 (15), 245 (22). To our knowledge, this is the first reported instance of fernenol in ferns.

6 β ,22-Dihydroxyhopane. Mp 238–242° (MeOH), $[\alpha]_D + 19^\circ$ (c 0.194). (Found: C, 81.09; H, 11.42. $C_{30}H_{52}O_2$ requires: C, 81.02; H, 11.79%.) IR $\nu_{max}^{CHCl_3}$ cm^{-1} : 3605 (OH). PMR δ : 0.79 (3H, s, C-28), 0.92 (3H, s, C-24), 0.96 (3H, s, C-27), 1.19 (3H, s, C-23), 1.21 (6H, s, C-25, C-26), 1.30 (6H, s, C-29, C-30),

4.55 (1H, m, $W_{1/2} = 10$ Hz, C-6). MS (70eV), m/e (rel. int.): 444 (M^+ , 4), 426 (3), 411 (1), 207 (21), 191 (16), 189 (52), 149 (34), 59 (100). Attempts to acetylate and oxidize **1a** under mild conditions were unsuccessful, giving unchanged starting material.

6 β , 22-diacetoxypopane (1b). Acetylation of **1a** by refluxing for 30 min. with isopropenyl acetate and *p*-toluenesulfonic acid as catalyst gave (**1b**): mp 146–150° (MeOH). IR $\nu_{max}^{CHCl_3}$ cm^{-1} : 1720, 1260 (OAc). PMR δ : 0.79 (3H, s, C-28), 0.94 (6H, s, C-24, C-27), 1.42 (3H, s, C-29 or C-30), 1.50 (3H, s, C-30 or C-29), 0.99, 1.15, 1.19 (9H, s, C-23, C-25, C-26 not respectively), 1.95 (3H, s, OAc), 2.02 (3H, s, OAc), 5.53 (1H, m, $W_{1/2} = 10$ Hz, C-6).

6 β ,22-Dihydroxyhopane (1a) from zeorin (1d). Zeorin was oxidized with CrO_3 -AcOH- H_2O to zeorinone (**1e**) [4] which was reduced with $LiAlH_4$ to epizeorin (**1a**) [5], identical to the natural product (**1a**) (mmp, TLC, IR, PMR).

Acknowledgement—We thank Prof. D. H. R. Barton for supplying us with a sample of zeorin.

REFERENCES

1. Corbett, R. E. and Young, H. (1966) *J. Chem. Soc. (c)* 1556.
2. Cheung, H. T. and Williamson, D. G. (1969) *Tetrahedron* **25**, 119.
3. Nishimoto, K., Ito, M., Natori, S. and Ohmoto, T. (1968) *Tetrahedron* 1043.
4. Barton, D. H. R. and Bruun, T. (1952) *J. Chem. Soc.* 1683.
5. Huneck, S. (1961) *Chem. Ber.* **94**, 1151.

Phytochemistry, 1976, Vol. 15, pp. 1997–1999. Pergamon Press. Printed in England.

6-METHOXYBENZOXAZOLINONE AND TRITERPENOID FROM ROOTS OF *SCOPARIA DULCIS*

CHIU-MING CHEN and MING-TYAN CHEN

Department of Chemistry, National Tsing Hua University, Hsinchu, Taiwan, China

(Received 24 June 1976)

Key Word Index—*Scoparia dulcis*; Scrophulariaceae; 6-methoxybenzoxazolinone; betulinic acid; ifflaionic acid.

INTRODUCTION

Scoparia dulcis L. (Scrophulariaceae), known as a folk-medicinal plant, has been used as a cure for diabetes mellitus in India and for hypertension in Taiwan [1]. To date, only common constituents; sitosterol, hexacosanol, D-mannitol, tritriacontane [2,3] and some unidentified compounds: dulciol, scopanol, dulciolone [3] and an alkaloid [4] have been isolated from this plant. The biologically active substance has been found, although the Indian drug "amellin" [5], which was obtained from the fresh plant, has been used as the antidiabetic principle [6]. The present report deals with the isolation of three compounds; 6-methoxybenzoxazolinone, betulinic acid and ifflaionic acid from the plant material which was collected at Hsinchu, Taiwan and identified at the Herbarium of National Taiwan University.

RESULTS AND DISCUSSIONS

Dried powdered root of *Scoparia dulcis* was extracted with Et₂O. Subsequent separation of the Et₂O extract

was achieved by repeated column chromatography using Si gel to give three crystalline compounds designated as compound A, B and C respectively.

Compound A (**1**), colourless needles from Me₂CO-petrol, mp 157–8°, IR $\nu_{max}^{cm^{-1}}$: 3350–3000 (N-H), 1770 (C=O, carbamate), 1620, 1490 (C=C, aromatic) and 1300 (C-N), UV λ_{max}^{nm} (log ϵ): 209(4.01), 232(4.05) and 290 (3.80), had the molecular formula $C_8H_7NO_3$, as indicated by its molecular ion peak at m/e 165 and the elemental analysis. The NMR spectrum of compound A in $(CD_3)_2CO$ showed a three-protons singlet at 3.75 ppm corresponding to a methoxy group, and three aromatic protons at 6.62 (*dd*, $J_1 = 9$, $J_2 = 3$ Hz), 6.8 (*d*, $J = 3$ Hz) and 6.93 ppm (*d*, $J = 9$ Hz) indicating the presence of the 1,2,4-trisubstituted benzene ring. Compound A could be acetylated with Ac₂O-Py to give an acetyl derivative (**2**), mp 151–2° and methylated with CH₂N₂ to give a N-methyl derivative (**3**), mp 97–9°. More detailed NMR studies led to the structure of **1** as 6-methoxybenzoxazolinone which was isolated from maize and wheat plants [7]. This identification was confirmed by direct compari-